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## Structures of Colchicine Analogues. V. 2-Methoxy-5-(4-methoxyphenyl)cyclohepta-2,4,6-trien-1-one

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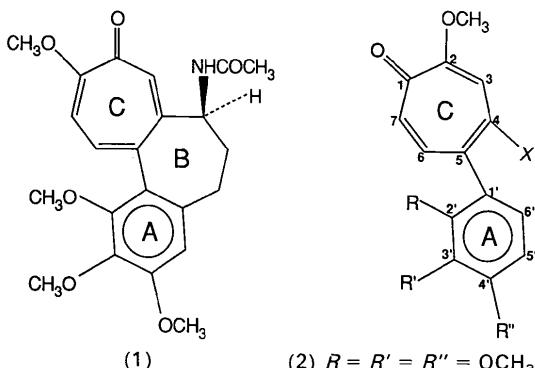
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**Abstract.**  $C_{15}H_{14}O_3$ ,  $M_r = 242.3$ , monoclinic,  $Pc$ ,  $a = 13.219(1)$ ,  $b = 7.072(1)$ ,  $c = 6.527(2)\text{ \AA}$ ,  $\beta = 97.23(1)^\circ$ ,  $V = 605.3(3)\text{ \AA}^3$ ,  $Z = 2$ ,  $D_m$  (flotation) = 1.33(1),  $D_x = 1.329\text{ Mg m}^{-3}$ ,  $\lambda(\text{Cu } K\alpha) = 1.5418\text{ \AA}$ ,  $\mu(\text{Cu } K\alpha) = 0.66\text{ mm}^{-1}$ ,  $F(000) = 256$ ,  $T = 291(1)\text{ K}$ , final  $R = 0.040$  for 905 observed data. The bicyclic molecule adopts a conformation similar to the solid-state conformation of isocolchicine, and the dihedral angle between the planes of the two rings is 41.1(4)°.

of colchicine analogues which might have potential as antimitotic agents (Banwell, Gravatt, Buckleton, Clark & Rickard, 1989; Banwell, Collis, Crisp, Lambert, Reum, Scoble, Gable, Mackay & Hamel, 1991), we report here the structure of an AC-ring mono-methoxyphenyl analogue (4). In earlier studies we reported the structures of the three bicyclic dimethoxyphenyl analogues (5) and (6) (Gable, Mackay, Banwell & Lambert, 1990) and (7) (Banwell *et al.*, 1991).



- (2)  $R = R' = R'' = \text{OCH}_3$ ,  $X = \text{H}$   
 (3)  $R = R' = R'' = \text{OCH}_3$ ,  $X = \text{Cl}$   
 (4)  $R = R' = X = \text{H}$ ,  $R'' = \text{OCH}_3$   
 (5)  $R = R' = \text{OCH}_3$ ,  $R'' = X = \text{H}$   
 (6)  $R' = R'' = \text{OCH}_3$ ,  $R = X = \text{H}$   
 (7)  $R = R'' = \text{OCH}_3$ ,  $R' = X = \text{H}$

**Experimental.** Pale-yellow prismatic crystals of (4) from benzene, *ca*  $0.24 \times 0.21 \times 0.27$  mm, aligned on a Rigaku-AFC diffractometer; cell parameters determined by least-squares from  $2\theta$  values for 25 strong reflections ( $41 < 2\theta < 78^\circ$ ); Cu  $K\alpha$  radiation (graphite crystal monochromator,  $\lambda = 1.5418$  Å);  $\omega-2\theta$  scan, scan rate  $2^\circ \text{ min}^{-1}$ , scan range ( $\Delta\omega$ )  $1.2^\circ + 0.5^\circ \tan\theta$ ,  $2\theta_{\max} = 130^\circ$ , 10 s stationary background counts; three standard reflections monitored every 50 reflections, no significant intensity variation; 1109 unique data,  $h$  0 to 15,  $k$  0 to 8,  $l$  -7 to 7, 905 for which  $I > 2\sigma(I)$  used for refinement; intensities corrected for Lorentz and polarization effects and for absorption, transmission factors 0.959 to 0.870. Structure solved by direct methods with *SHELXS86* (Sheldrick, 1985) and refined with *SHELX76* (Sheldrick, 1976). H-atom sites located by difference, except for those of the methoxy groups, which were included at idealized positions (C—H 1.08 Å); no refinement of H-atom coordinates or temperature factors; full-matrix least-squares refinement with anisotropic temperature factors given to C and O atoms, isotropic for H, converged at  $R = 0.040$  and  $wR = 0.052$ ,  $S = 1.27$  (162 parameters varied); function minimized  $\sum w(|F_o| - |F_c|)^2$ , with weights  $(\sigma^2|F_o| + 0.0012|F_o|^2)^{-1}$ ; one reflection, 020, apparently seriously affected by extinction omitted from the final refinement. At convergence,  $(\Delta/\sigma)_{\max} = 0.001$ ;  $(\Delta\rho)_{\max}, (\Delta\rho)_{\min} = +0.14, -0.21 \text{ e } \text{\AA}^{-3}$ . Atomic scattering factors and anomalous-dispersion factors from *International Tables for X-ray Crystallography* (1974, Vol. IV, pp. 99, 149). Figures were prepared from the output of *ORTEPII* (Johnson, 1976). Calculations performed on a VAX11/780 computer.

**Discussion.** Final atomic coordinates for the non-H atoms are given in Table 1,\* bond lengths and angles are listed in Table 2 and the molecular conformation and numbering is illustrated in Fig. 1. The molecule (4) adopts a solid-state conformation that resembles isocolchicine (Lessinger & Margulis, 1978a), an inactive isomer of colchicine, rather than colchicine (Lessinger & Margulis, 1978b) (1). This solid-state conformation has been adopted without exception by all other bicyclic *AC*-ring analogues so far studied, namely (2) (Rossi, Link & Lee, 1984), (3) (Banwell *et al.*, 1988), (5) and (6) (Gable *et al.*, 1990), and (7) (Banwell *et al.*, 1991). The tropolone *A*-ring atoms in (4) are coplanar to within 0.04 (1) Å, and

Table 1. *Final atomic coordinates ( $\times 10^4$ ) and equivalent isotropic temperature factors for the non-H atoms with e.s.d.'s in parentheses*

	$x$	$y$	$z$	$B_{\text{eq}}$ (Å $^2$ )
C(1)	4196	2848 (5)	4979	3.26 (7)
C(2)	4501 (4)	2364 (5)	2932 (8)	3.25 (7)
C(3)	3862 (4)	1968 (5)	1162 (8)	3.17 (6)
C(4)	2783 (4)	1930 (5)	752 (9)	2.99 (6)
C(5)	2043 (5)	2362 (5)	1950 (9)	2.86 (6)
C(6)	2253 (4)	2912 (5)	4066 (8)	2.95 (6)
C(7)	3162 (4)	3144 (5)	5316 (6)	3.30 (6)
O(1)	4872 (4)	3057 (6)	6456 (6)	5.62 (7)
O(2)	5516 (4)	2333 (4)	3034 (6)	3.95 (5)
C(2')	5976 (5)	1956 (8)	1203 (9)	5.46 (10)
C(1')	950 (4)	2359 (4)	1017 (9)	2.71 (6)
C(2'')	667 (4)	3099 (5)	-940 (8)	3.13 (6)
C(3'')	-349 (4)	3148 (5)	-1855 (8)	3.30 (7)
C(4'')	-1093 (5)	2427 (5)	-756 (9)	3.20 (6)
C(5'')	-821 (4)	1683 (5)	1214 (9)	3.39 (6)
C(6'')	182 (4)	1674 (5)	2084 (8)	3.32 (6)
O(4'')	-2115 (4)	2364 (4)	-1491 (9)	4.35 (6)
C(4''')	-2441 (5)	3158 (8)	-3452 (10)	5.35 (10)

Table 2. *Bond lengths (Å) and angles (°) with e.s.d.'s in parentheses*

C(1)—C(2)	1.483 (5)	O(2)—C(2'')	1.432 (8)
C(1)—C(7)	1.427 (5)	C(1')—C(2'')	1.387 (7)
C(1)—O(1)	1.238 (5)	C(1')—C(6'')	1.389 (8)
C(2)—C(3)	1.371 (7)	C(2'')—C(3'')	1.400 (7)
C(2)—O(2)	1.335 (7)	C(3'')—C(4'')	1.385 (8)
C(3)—C(4)	1.418 (7)	C(4'')—C(5'')	1.394 (8)
C(4)—C(5)	1.362 (9)	C(4'')—O(4'')	1.376 (8)
C(5)—C(6)	1.428 (8)	C(5'')—C(6'')	1.375 (7)
C(5)—C(1')	1.496 (8)	O(4'')—C(4'')	1.414 (8)
C(6)—C(7)	1.375 (7)		
C(2)—C(1)—C(7)	123.2 (5)	C(2)—O(2)—C(2'')	119.4 (4)
C(2)—C(1)—O(1)	118.5 (3)	C(5)—C(1')—C(2'')	120.6 (4)
C(7)—C(1)—O(1)	118.3 (3)	C(5)—C(1')—C(6'')	121.7 (4)
C(1)—C(2)—C(3)	126.7 (3)	C(2'')—C(1')—C(6'')	117.6 (4)
C(1)—C(2)—O(2)	110.1 (3)	C(1')—C(2'')—C(3'')	122.4 (4)
C(3)—C(2)—O(2)	123.2 (4)	C(2'')—C(3'')—C(4'')	118.4 (4)
C(2)—C(3)—C(4)	131.4 (4)	C(3'')—C(4'')—C(5'')	119.9 (4)
C(3)—C(4)—C(5)	131.7 (4)	C(3'')—C(4'')—O(4'')	124.1 (4)
C(4)—C(5)—C(6)	123.4 (4)	C(5'')—C(4'')—O(4'')	115.9 (4)
C(4)—C(5)—C(1')	119.5 (4)	C(4'')—C(5'')—C(6'')	120.4 (4)
C(6)—C(5)—C(1')	117.0 (4)	C(1')—C(6'')—C(5'')	121.3 (4)
C(5)—C(6)—C(7)	131.0 (4)	C(4'')—O(4'')—C(4'')	118.5 (4)
C(1)—C(7)—C(6)	132.4 (3)		

both methoxy groups are coplanar with their associated rings as reflected by the torsion angles C(1)—C(2)—O(2)—C(2'') of  $-177.9 (4)^\circ$  and C(3')—C(4')—O(4')—C(4'') of  $-3.0 (8)^\circ$ . The dihedral angle between the normals to the two rings in the other bicyclic analogues (see scheme) ranges between  $43.4 (3)^\circ$  in (6) and  $57.4 (5)^\circ$  in (2).

As only hydrogen-bond acceptors are present in the analogue molecules and the crystals are not solvated, hydrogen-bond formation is precluded. Thus the molecules are held together in the crystal by van der Waals interactions only, with the shortest intermolecular contacts being C(2'')···O(4')(1 +  $x, y,$

\* Lists of structure factors and H-atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 54562 (12 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: HA0070]

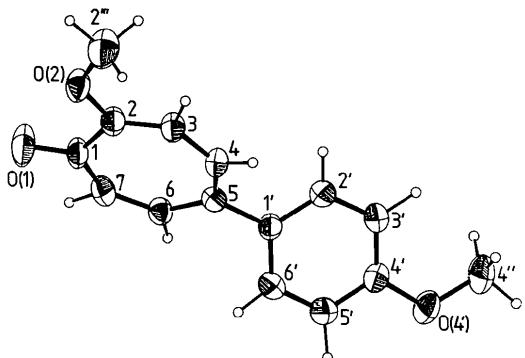


Fig. 1. Perspective view of (4) with thermal ellipsoids scaled to 50% probability. H atoms are denoted by spheres of arbitrary radius.

$z)$  3.265 (9) Å, and C(2'')···O(1)( $x$ ,  $y$ ,  $-1+z$ ) 3.346 (7) Å.

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## Structure of a Substituted 2-Thiohydantoin

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**Abstract.** *S*-[1-(3-Acetyl-5-oxo-2-thioxo-2,3,4,5-tetrahydro-1*H*-imidazol-4-yl)ethyl] ethanethioate, C<sub>9</sub>H<sub>12</sub>N<sub>2</sub>O<sub>3</sub>S<sub>2</sub>,  $M_r = 260.3$ , monoclinic,  $P2_1/n$ ,  $a = 8.643$  (1),  $b = 15.554$  (1),  $c = 8.898$  (1) Å,  $\beta = 92.05$  (1)°,  $V = 1195.4$  (3) Å<sup>3</sup>,  $Z = 4$ ,  $D_m$ (flootation) = 1.448 (5),  $D_x = 1.446$  Mg m<sup>-3</sup>,  $\lambda(\text{Cu } K\alpha) = 1.5418$  Å,  $\mu(\text{Cu } K\alpha) = 3.96$  mm<sup>-1</sup>,  $F(000) = 544$ ,  $T = 293$  (1) K, final  $R = 0.046$  for 1708 observed data. Atoms of the thiohydantoin nucleus are approximately coplanar, and the N(3) acetyl group is twisted by about 12° from the mean plane. N(1) of the

hydantoin ring is the donor atom in an intermolecular hydrogen bond with the carbonyl oxygen of the N(3) acetyl substituent, the N(1)···O(6) distance being 2.873 (3) Å. These interactions link the molecules into chains along the [101] direction in the crystal.

**Introduction.** Application of the thiocyanate degradation procedure (Schlack & Kumpf, 1926) to a peptide (I) converts the C-terminal amino acid into a substituted thiohydantoin derivative (II), which can